

# User-friendly novel method of geopolymer concrete production mixes

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## Abstract

Geopolymer is an inorganic polymer, which can perform as an effective binder in place of Portland cement. With this, the Geopolymer concrete has become highly Ecofriendly with very low carbon footprint. The GPC formulation at present utilises industrial waste materials like fly ash and GGBS, which are activated with well-proportioned alkaline liquid. At present the commercially available lye and Sodium Silicate solution are used to prepare alkaline liquid capable of initiating geopolymer reaction. It is noticed that the liquid system has limited shelf life by making field production of Geopolymer concrete mixes was a difficult one more over the liquid system and its ingredients are difficult to transport over longer distances. This has been the main obstacle for development of large-scale applications of GPC mixes on commercial basis. This issue was addressed in the present study by using solid activator in replacement of liquid solution. The solid activating system can be produced just before the preparation of fresh GPC mixes at anyway at any time. It is found that the mechanical strength (40 – 45Mpa) property was similar to the liquid activating GPC mixes. This new technique of production of GPC mixes using solid activators can be consider as the major breakthrough for commercialization / large scale field applications of Geopolymer technology.

**Keywords:** Geopolymer; Fly ASH; GGBS; Compressive Strength

## 1. Introduction

Geopolymer term was initially coined by French material scientist J. Davidovits in the year 1979. Geopolymer binder is an innovative and eco-friendly construction material and an alternative to Portland cement binder. As per the latest report by National Control of Pollution Board, the Ordinary Portland cement emits about 868 million tonnes of CO<sub>2</sub>. So in order to reduce the emission of CO<sub>2</sub> the idea of using geopolymer concrete was done. Both Fly ash and GGBS are processed by appropriate technology and used for concrete works reducing the emission of CO<sub>2</sub> by Ordinary Portland Concrete. The geopolymer concrete is composed of fly ash, GGBS (Granulated blast furnace slag) and an Alkaline Activator solution, which is mainly made up of liquid sodium silicate solution available commercially in the market. It is one of the major constituents of geopolymer source material since it activates the geopolymeric source materials containing silica and aluminium such as Fly ash and GGBS. AAS (alkaline activator solution) is mainly used as liquid binder in the concrete is also called as RGL (Reaction Generating Liquid). Geopolymer can be considered as an eco-friendly concrete since it uses the waste materials emissions from coal power plants that is flyash and GGBS which is a residue of ceramic industries, therefore a sustainable environment is created by recycling the waste products from the following factories. Geopolymer concrete is also an self curing concrete since it requires no water and can be demoulded within 3 to 4 hrs after casting.

In this paper the prime focus is on replacing the Liquid sodium silicate solution by Solid sodium silicate. Since the transportation and handling of liquid solution to the site is a factor of major concern plan has been made to make the solid sodium silicate solution

in site itself and use it for geopolymer concrete production. Recent literature studies done by “FN Okoye[6]” reveals that silicates increases the compressive strength. Previous studies has been done with the help of liquid silicate solutions by “E. Kamseui, Mandeepkaur Jaswal[9], [10]” so the idea is to replace the liquid silicate with solid sodium silicate solution and compare the compressive strengths. In the following paper we will discuss about the use of sodium metasilicate with different grades of Na<sub>2</sub>SiO<sub>3</sub>.5H<sub>2</sub>O (RGL3), 7H<sub>2</sub>O (RGL2) and 9H<sub>2</sub>O (RGL1) and compare the strength it has obtained with that of normal RGL. RGL 3 has been considered to be superior than other two RGLs and its compressive strength is almost similar to that of normal RGL. It also discusses about strength change with different mix proportions of Flyash and GGBS used in RGL3. The following project will be a good breakthrough in the field of Geopolymer concrete production and it will be economical as there is no requirement of liquid silicate solution transportation.

The major objective of the paper is to revolutionize the geopolymer concrete industry by replacing the liquid silicate solution with solid sodium silicate by making it on the site, making the production of geopolymer concrete economical and viable. The transportation cost for solution will be significantly reduced. Thus it will make the production of geopolymer concrete more bulkier and economical.

## 2. Materials and mix design

### 2.1. Materials

Geopolymeric source material used in the present work is Class C fly ash from Ennore thermal power plant, India and GGBS from the Jindal power plant. The fine aggregate used is river sand has been used here as a fine aggregate with a sieve of 4.75mm. The coarse aggregates used for casting purposes are of size 10mm and 12mm.

## 2.2. Reaction generating liquid

Reaction generating liquid is traditionally prepared by mixing the sodium silicate solution (MR: 2.2) with NaOH (50% lye) and it is denoted as RGL.

A novel method of preparing RGL by solid sodium silicate powder as follows:

Sodium metasilicate mixed with distilled water of grades  $\text{Na}_2\text{SiO}_3 \cdot 9\text{H}_2\text{O}$  (RGL1),  $\text{Na}_2\text{SiO}_3 \cdot 7\text{H}_2\text{O}$  (RGL 2) and  $\text{Na}_2\text{SiO}_3 \cdot 5\text{H}_2\text{O}$  (RGL 3) followed by the NaOH addition respectively. The Physico-chemical properties of the RGL are listed in table.1 mentioned below:

**Table1:** Physico-Chemical Properties of RGL

S.no	Composition	Physical properties		Chemical Composition		
		Appearance	Shape	$\text{Na}_2\text{O}_3$	$\text{SiO}_2$	$\text{H}_2\text{O}$
1.	$\text{Na}_2\text{SiO}_3 \cdot 9\text{H}_2\text{O}$	White Shape	Crystalline	22%	22%	66%
2	$\text{Na}_2\text{SiO}_3 \cdot 7\text{H}_2\text{O}$	Cream Colour Powder	Crystalline	24%	25%	51%
3	$\text{Na}_2\text{SiO}_3 \cdot 5\text{H}_2\text{O}$	White Colour Powder	Crystalline	28%	29%	43%
4	$\text{Na}_2\text{SiO}_3$	White transparent fluid	Crystalline	15%	30%	55%

## 2.3. Mix design

Fly Ash/GGBS( $\text{kg}/\text{m}^3$ )	Fine Aggregate( $\text{kg}/\text{m}^3$ )	Coarse Aggregate( $\text{kg}/\text{m}^3$ )	RGL( $\text{kg}/\text{m}^3$ )	L/S ratio
407	647	1056	233	0.55

## 3. Experimental programme

This paper discusses about the various experimental works that has been done with the use of different RGLs (RGL1, RGL2 and RGL3) and comparing the strength with normal RGL. Initially the RGL mix was prepared of mix ratio 1:4:10 which composes of 1 lye, 4 metasilicate and 10 distilled water. The various specimens has been casted with different RGLs of same mix and the strength was compared. The RGL3 of grade 5  $\text{H}_2\text{O}$  was selected among the following RGLs considering its availability factor and strength. In the next phase the specimens were casted using RGL 3 with different mix proportions of fly ash and GGBS in the ratio of 50:50 and 80:20. Change in strength was studied with the mix proportions of 50:50 and 80:20. Various specimens were casted using the mix proportions of 50:50 and 80:20 and its properties were studied. The strength was calculated for 3, 7 and 28 days. Earlier studies were done by "G.Naidu, V.Prasad [19]" to improve the strength properties.

The normal RGL used is of ratio 1:2:7. RGL 1, 2 and 3 was prepared in the lab. RGL was prepared in the mix ratio of 1:4:10, which means 1 sodium hydroxide flakes, 4 water and 10  $\text{H}_2\text{O}$ . The liquid solution is having a molar ratio of 2.5. The compressive strength test was done for the specimens and its strength were compared. Earlier powder preparation was done by using inorganic cement based activates slag was done by "Kaituo Wang [4]". He has discussed the use of solid watch glass for preparation of RGL solution.

## 4. Results and discussion

### 4.1. Characterisation of RGL

#### 4.1.1. Temperature and solubility test

Initially solubility test was done for the various RGLs. It was done by initially taking metasilicates and dissolving it in a certain quantity of water. The metasilicates was added slowly and the solution was stirred properly so that metasilicate gets completely dissolved in water and after 2 to 3 minutes sodium hydroxide flakes was added in the water. During its addition a certain amount of heat was liberated thereby producing exothermic reaction. The solution was stirred properly until it gets a clear solution. In case of RGL 2 it was observed that solution was not getting dissolved to a certain extent and some materials were getting precipitated at bottom. The temperature was noted during the addition of flakes. To improve the polymerisation reaction in sodium silicate solution, high temperature is required as discussed in "[2-4]".

The temperature tests of various solutions has been observed when the flakes were added into it. The temperature was noted for the solutions with grade composition of  $5\text{H}_2\text{O}$ ,  $7\text{H}_2\text{O}$  and  $9\text{H}_2\text{O}$ . First of all the temperature was noted for the mix with powder composition of  $9\text{H}_2\text{O}$ . Initially the room temperature was noted. The room temperature before adding the flakes was  $33^\circ\text{C}$ . After adding the flakes the temperature drastically increased by 4 to 5 degrees. The temperature of the solution was measured using infrared thermometer. It was observed that there is an increase of 4 to 5 degree more than room temperature after adding the flakes. Similarly the temperature test was done for 7  $\text{H}_2\text{O}$  and it was observed that there was only a marginal increase of 3 to 4 degrees only. After making the RGL mix with  $5\text{H}_2\text{O}$ , the temperature test was done after adding 200g flakes and 400 g flakes into it. It was observed that the temperature gone very high after adding the flakes in the mix. The initial room temperature was  $28^\circ\text{C}$  and after adding the flakes it was observed that the temperature was around  $45^\circ\text{C}$ . It is observed that there is a high increase in temperature with RGL mix containing  $5\text{H}_2\text{O}$ .

### 4.2. Compressive strength test results for RGLS

The Compressive strength test result has been done for specimens made of different RGLs. The compressive strength of the specimen was checked for 7, 14 and 28 days. The L/S ratio of specimen was 0.55 while for RGL 2 the L/S ratio was 0.6.

The 100mm cube specimens has been casted by using the different RGLs. The different parameters such as solubility and temperature were considered while casting.

#### 4.2.1. Workability of RGLs

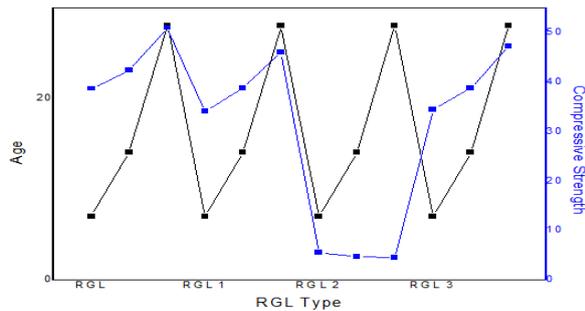
It has been observed that RGL 3 had been giving a better workability when compared to the normal RGL. The workability mix of RGL 1 and RGL 2 was good.

**Table 2:** Compressive Strength of Different Rgls

Mix ID	Age	Average Compressive Strength(Mpa)	L/S ratio
RGL	7	38.6	0.55
	14	42.3	
	28	50.86	
RGL 1	7	34	0.55
	14	38.7	
	28	46	
RGL 2	7	5.5	0.6
	14	4.7	
	28	4.4	
RGL 3	7	34.36	0.55
	14	38.73	
	28	47.2	

#### 4.2.2. Strength analysis

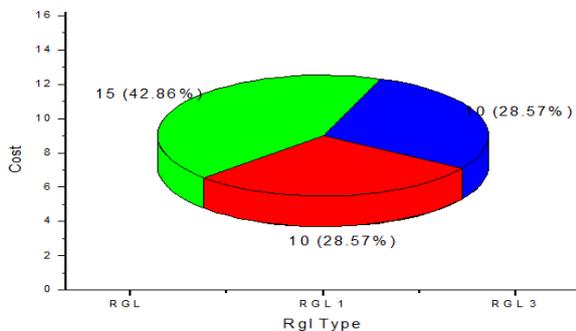
From the above results we can get a clear idea about the compressive strength of different RGLs. The calculation of compressive strength was done for specimen of 100mm cube size. There was a considerable strength reduction for RGL 2 while RGL 1 and RGL 3 was giving a fair strength when compared with that of RGL. RGL 1 is not available commercially in the market. The material available is of laboratory grade. While RGL 2 and RGL 3 available is of commercial grade. It was observed that strength of RGL 2 was considerably lower when compared to the other RGLs. Since strength reduction was higher, it was not a good idea to use RGL 2.



Compressive Strength Variations in Different RGLs.

#### 4.2.3. Cost analysis

Comparing the cost of different RGLs, RGL 3 was compared to be cheaper than the other RGLs. RGL is liquid solution and its transportation and handling is difficult compared to the other RGLs, while RGL 1 available is of laboratory grade and is not available commercially in the market. RGL 2 and RGL 3 are available commercially in the market but citing the strength reduction of RGL 2, RGL 3 is preferred. RGL 3 is available within the market commercially.



Cost analysis of various RGLs.

From the above results we can select the RGL 3 mix since it was easily available in the market. The strength of RGL 3 was better with respect to the other two RGLs. The various cylindrical specimens of 100mm dimension were casted using RGL 3 and its compressive strength was compared as that of normal RGL.

#### 4.2.4. Availability and convenience

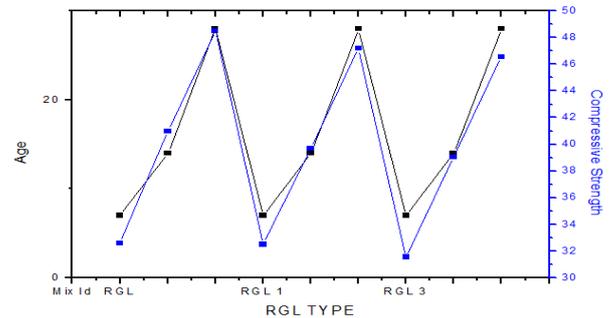
The RGL 3 solution is easily available and is convenient. The solution does not require proper handling and care since it is available in the solid form and there is easiness in transporting the solid material to site. Thus the transportation cost of the material can be reduced to a further extent based on its easy availability.

#### 4.2.5. Storage of materials

Since the material is available in solid form the storage of material is not a matter of concern while the liquid solution has to be stored in a container. The solution is prepared by using the raw materials

available commercially and is immediately mixed into the concrete.

#### 4.3. Compressive strength study of different RGLs on cylinder specimens (100mm)



#### 4.3.1. Characteristics of compressive strength with different RGLs on cylinders

From the graphical analysis it can be said that RGL 1 and RGL 3 have fair strength when compared to the normal RGL. The strength of around 47 Mpa is obtained after a period of 28 days. The L/S ratio was 0.5. The cylindrical specimens used for casting was of 100mm diameter.

### 5. Conclusion

The Compressive strength of different RGLs is studied from this paper. From the results available it indicates that there is a scope to use the solid sodium silicate instead of using the liquid silicate solution. Compressive strength of different RGLs indicate that RGL 3 has a better performance considering its strength, availability and cost factors. The strength of RGL 3 was almost similar to that of normal RGL. The RGL 3 solution is easy to make hence it is time saving. It easily binds the aluminosilicate particles. Thus with the help of available data we can say that RGL 3 can be used as a better alternative to that of normal RGL.

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